

Physical Constants

Acceleration of gravity at sea level and equator	$g = 9.806 \text{ m s}^{-2}$
Atomic mass unit	$1 \text{ amu} = 1.660566 \times 10^{-24} \text{ g} = (6.022045 \times 10^{23})^{-1} \text{ g}$
Avogadro's number	$N = 6.022045 \times 10^{23} \text{ mole}^{-1} \text{ or amu g}^{-1}$
Bohr radius	$a_o = 0.529177 \text{ \AA} = 5.29177 \times 10^{-11} \text{ m}$
Boltzmann's Constant	$k = 1.38066 \times 10^{-23} \text{ JK}^{-1} \text{ molecule}^{-1}$
Electron charge	$e = 1.602189 \times 10^{-19} \text{ C}$
Faraday's constant	$F = 96,484.6 \text{ C mole}^{-1}$
Gas constant	$R = 8.31441 \text{ J K}^{-1} \text{ mole}^{-1}$ $R = 0.082057 \text{ L atm K}^{-1} \text{ mole}^{-1}$
Mass of electron	$m_e = 9.109534 \times 10^{-31} \text{ kg} = 0.0005486 \text{ amu}$
Mass of proton	$m_p = 1.672649 \times 10^{-27} \text{ kg} = 1.007276 \text{ amu}$
Mass of neutron	$m_n = 1.674954 \times 10^{-27} \text{ kg} = 1.008665 \text{ amu}$
Mass of proton plus electron	$m_{p+e} = 1.673560 \times 10^{-27} \text{ kg}$
Planck's constant	$h = 6.626176 \times 10^{-34} \text{ J s}$ $\frac{h}{2\pi} = \hbar = 1.054589 \times 10^{-34} \text{ J s}$
Rydberg Constant	$R_\infty = 109,737.32 \text{ cm}^{-1}$
Speed of light	$c = 2.9979246 \times 10^8 \text{ m s}^{-1}$
Mass of an alpha particle	$6.6 \times 10^{-27} \text{ kg}$

Conversion Factors

1 atm = 101.3 kPa = 760 mm Hg = 760 torr
0°C = 273.15 K
$x^\circ\text{F} = (9^\circ\text{F}/5^\circ\text{C}) y^\circ\text{C} + 32^\circ\text{F}$
1 electron volt (eV) = $1.602189 \times 10^{-19} \text{ J}$
1 calorie = 4.184 J
1 liter atm = 101.325 J
$2.303 RT = 5.7061 \text{ kJ mol}^{-1}$ at 298 K
$1 \text{ kJ mole}^{-1} = 83.593 \text{ cm}^{-1}$ (from $E = hc\bar{\nu}$)
$1 \text{ amu} = 1.49244 \times 10^{-10} \text{ J} = 931.502 \text{ MeV}$
1 curie (Ci) = 3.7×10^{10} Becquerel (Bq) = 3.7×10^{10} disintegrations s^{-1}

SI Scale Prefixes

Tera(T)	10^{12}	Kilo (k)	10^3	Milli (m)	10^{-3}
Giga(G)	10^9	Deci (d)	10^{-1}	Micro (μ)	10^{-6}
Mega(M)	10^6	Centi (c)	10^{-2}	Nano (n)	10^{-9}
				Pico (p)	10^{-12}

Fundamental Units

1 Joule = 1 J = $1 \text{ m}^2 \text{ kg s}^{-2}$
1 Newton = 1 N = 1 m kg s^{-2}
1 Pascal = 1 Pa = $1 \text{ N m}^{-2} = 1 \text{ J m}^{-3} = 1 \text{ kg m}^{-1} \text{ s}^{-2}$

Formulae

Properties of Light: $E = hf = hc/\lambda$, $c = f\lambda$

DeBroglie Wavelength: $\lambda = h/p$

Ideal Gases: $PV = nRT$, $\rho = \frac{P(\text{molar mass})}{RT}$

Geometric Formulae

Circumference of a Circle = $2\pi r$

Area of a Circle = πr^2

Volume of a Sphere = $\frac{4}{3}\pi r^3$

Surface area of a Sphere = $4\pi r^2$

Differentials

$$\frac{dx}{dx} = 1$$

$$\frac{dx^n}{dx} = nx^{n-1}$$

$$\frac{d \ln x}{dx} = 1/x$$

$$\frac{df(x)}{dy} = \frac{df(x)}{dx} \frac{dx}{dy}$$

Quadratic Formula

if $ax^2 + bx + c = 0$ then

Integrals

$$\int_{x_1}^{x_2} f(x)dx = - \int_{x_2}^{x_1} f(x)dx$$

$$\int_{x_1}^{x_2} dx = x_2 - x_1$$

$$\int_{x_1}^{x_2} x^n dx = \frac{x_2^{n+1} - x_1^{n+1}}{n+1}$$

$$\int_{x_1}^{x_2} \frac{dx}{x} = \ln \frac{x_2}{x_1}$$

$$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$$