

## ASSIGNMENT 6

### CHEMISTRY 300

Due: 4:30 pm Friday 3 November 2006

1. Fick's laws describes diffusion from at point source of  $N_0$  molecules:

$$n(r, t) = \frac{N_0 e^{-r^2/4Dt}}{(4\pi Dt)^{3/2}}$$

Bombykol ( $C_{16}H_{29}OH$ ) is a silkworm moth attractant pheromone. As few as 200 bombykol molecules in a cubic centimeter causes the male silkworm to flutter his wings in a characteristic frenzy. If 1.0 mg of bombykol is placed in a small spot 1.0 m from a male moth, when will the male moth go into a frenzy? The diffusion coefficient is  $D = 4.1 \times 10^{-7} \text{ m}^2 \text{ s}^{-1}$ .

2. If a  $1.00 \text{ cm}^2 \times 10.0 \text{ cm}$  bar of Cu ( $k=400 \text{ W m}^{-1} \text{ K}^{-1}$ ) has one end in contact with a 1.0 g sample of ice at  $0^\circ\text{C}$  and the other in contact with a hot plate at  $140^\circ\text{C}$ , how long does it take to melt the ice? The enthalpy of fusion for water is  $\Delta H_{fus}^\circ = 6.008 \text{ kJ mol}^{-1}$ .

3. The Poiseuille equation for an incompressible fluid is related to viscosity:

$$\frac{dV}{dt} = \frac{\pi R^4}{8L\eta} \Delta p$$

where  $R$  is the diameter and  $L$  is the length of a cylindrical pipe, and  $p$  is pressure.

Consider two copper pipes, each 3 m long, the first having 2.6 cm and the second having 1.3 cm inner diameter, are connected in series. A pressure of 5 atm is supplied at the opening of the wider pipe and oil ( $\eta = 0.114 \text{ Pa s}$ ) exits from the narrow end at a pressure of 1 atm.

- (a) Calculate the pressure at the point where the two pipes are joined.
  - (b) How many litres per minute of oil can be delivered by the combination.
4. Consider problem 3 in assignment 5. Derive an expression for  $(\kappa/d)_{eff}$  such that:

$$\Gamma_q = - \left( \frac{\kappa}{d} \right)_{eff} (T_{indoor} - T_{outdoor})$$