

ASSIGNMENT 4

CHEMISTRY 300

Due: 4:30 pm Wednesday 11 October 2006

1. Do question 6 on page 852 of the text.
2. Do question 8 on page 852 of the text.
3. Consider 200 cylindrical individuals in a mosh pit. The area of the mosh pit is $20\text{ m} \times 20\text{ m}$. The diameter of each individual is 0.30 m and their speed is 0.46 m s^{-1} . Assuming chaotic motion, calculate:
 - (a) the mean free path.
 - (b) the number of collisions per minute an individual is involved in.
 - (c) the total number of collisions per minute.
4. Consider a gas at constant temperature. If the pressure is doubled, what effect does this have on:
 - (a) the number of collisions per second made by any one molecule.
 - (b) the total number of collisions per second occurring in 1 m^3 of gas.
 - (c) the mean free path of a gas molecule.